



# Production of Hydrogen from Natural Gas with Integrated CO<sub>2</sub>-capture

## Fuel for a Thermal Power Plant

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## Objective of the present study:

**Suggest and describe how the "IFE" CO<sub>2</sub>-removal process**

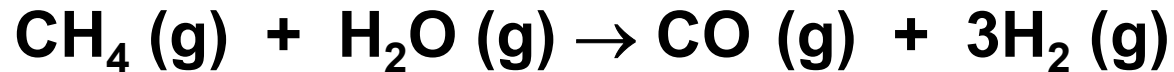


**can be integrated in a hydrogen Combined Cycle Power Plant (CCPP) and a hydrogen steam boiler**

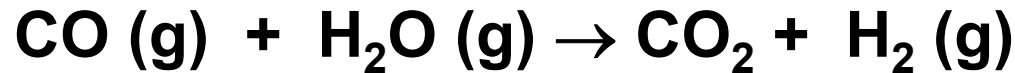
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- **Generate a N<sub>2</sub>-diluted, H<sub>2</sub> fuel gas stream for gas turbines fuel (H<sub>2</sub>/N<sub>2</sub>-ratio: 50/50)**
  - **Generate a H<sub>2</sub>-rich fuel (+ 95%) for steam boilers**

# H<sub>2</sub>-production in one single step

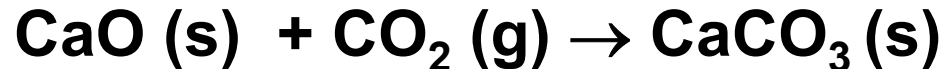
## Steam reforming:



## Water gas shift:



## Carbonation:

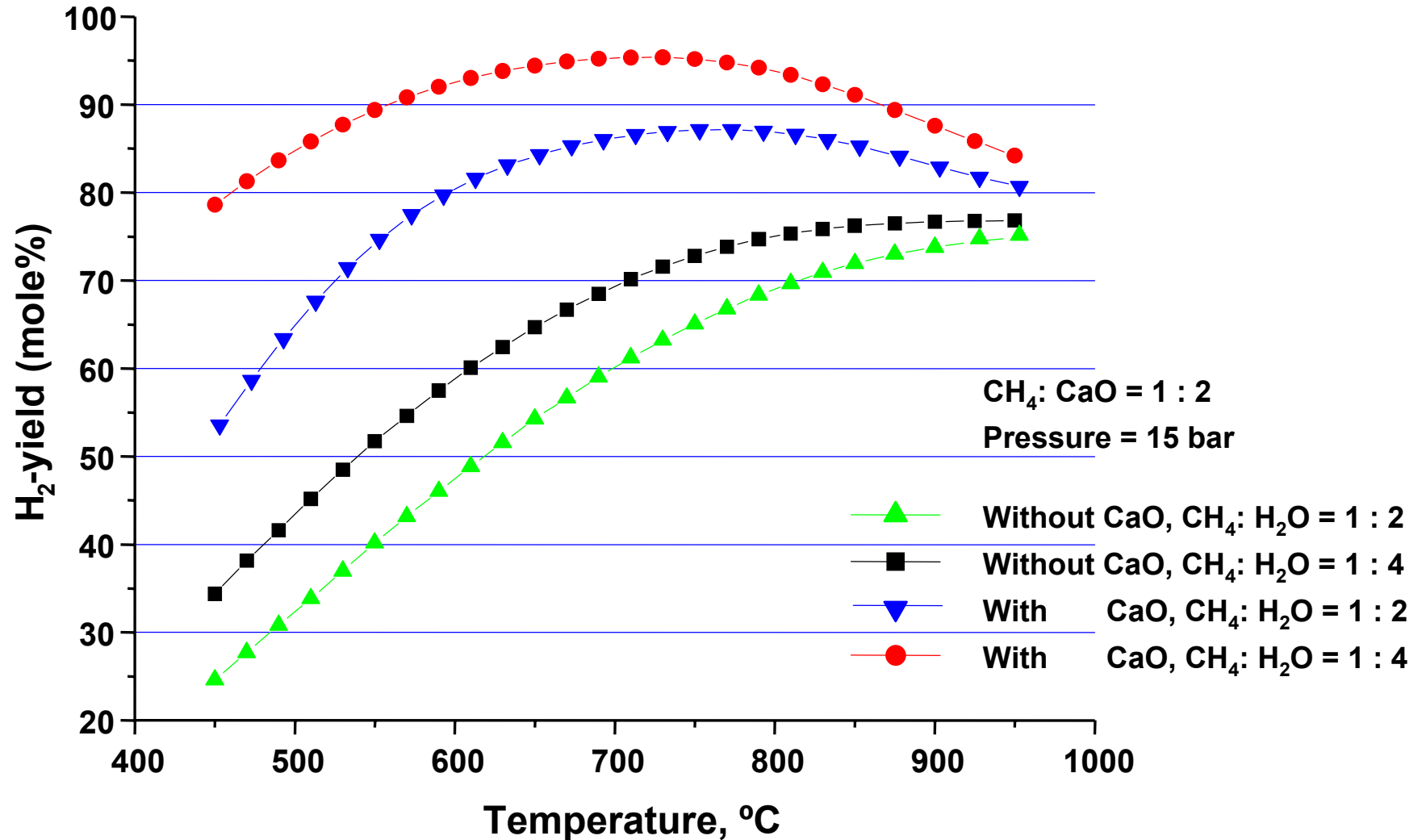


## Overall:



# Equilibrium hydrogen content

– thermodynamic analysis



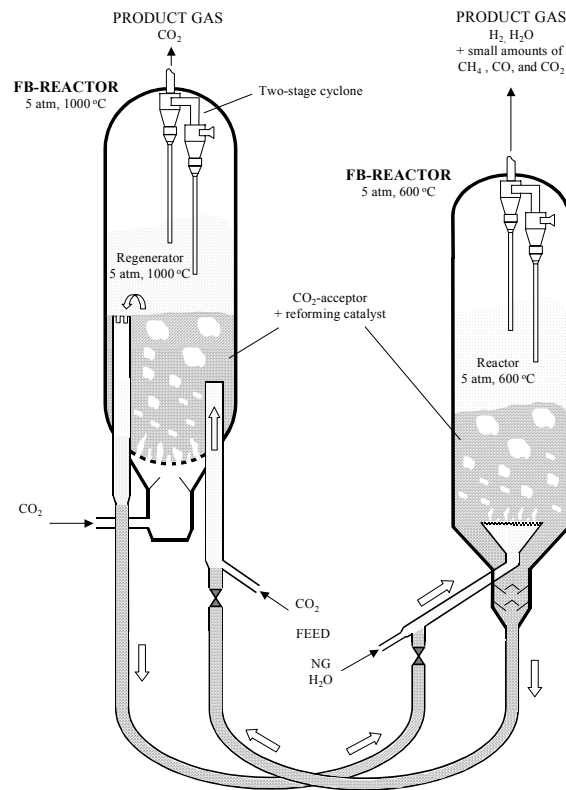
# Integrated reforming and CO<sub>2</sub> capture

## Advantages:

- **Process simplification**
  - Reforming, water gas shift and CO<sub>2</sub>-separation occur simultaneously in the same reactor
- **Increased hydrogen yield**
  - High H<sub>2</sub>-yield at lower temperatures than in the conventional reforming process
- **Separation of CO<sub>2</sub> as a solid in the process**
  - No additional costly step for CO<sub>2</sub> separation
  - Delivered as a pressurised, concentrated CO<sub>2</sub> stream for sequestration or utilisation
- **Recycled CaO represents an important carrier of heat into the reforming stage**

# Challenge: Reactor technology -

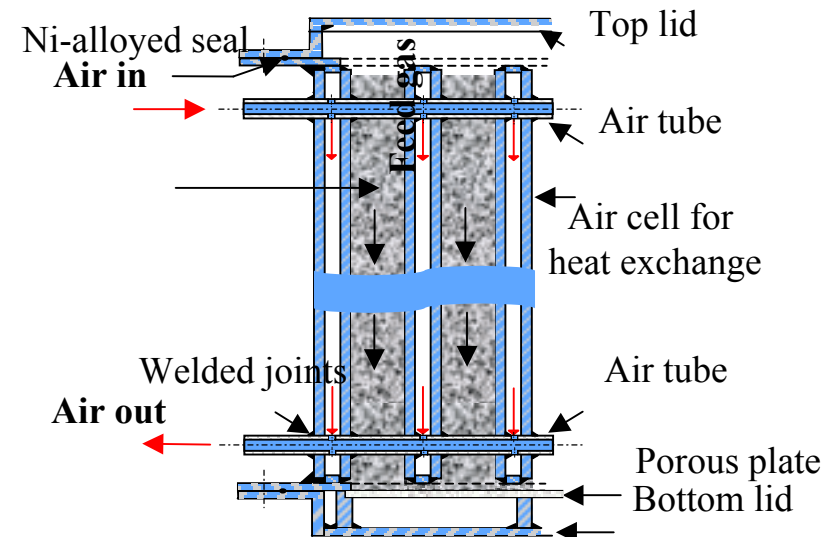
## Continuous process; circulation of solids

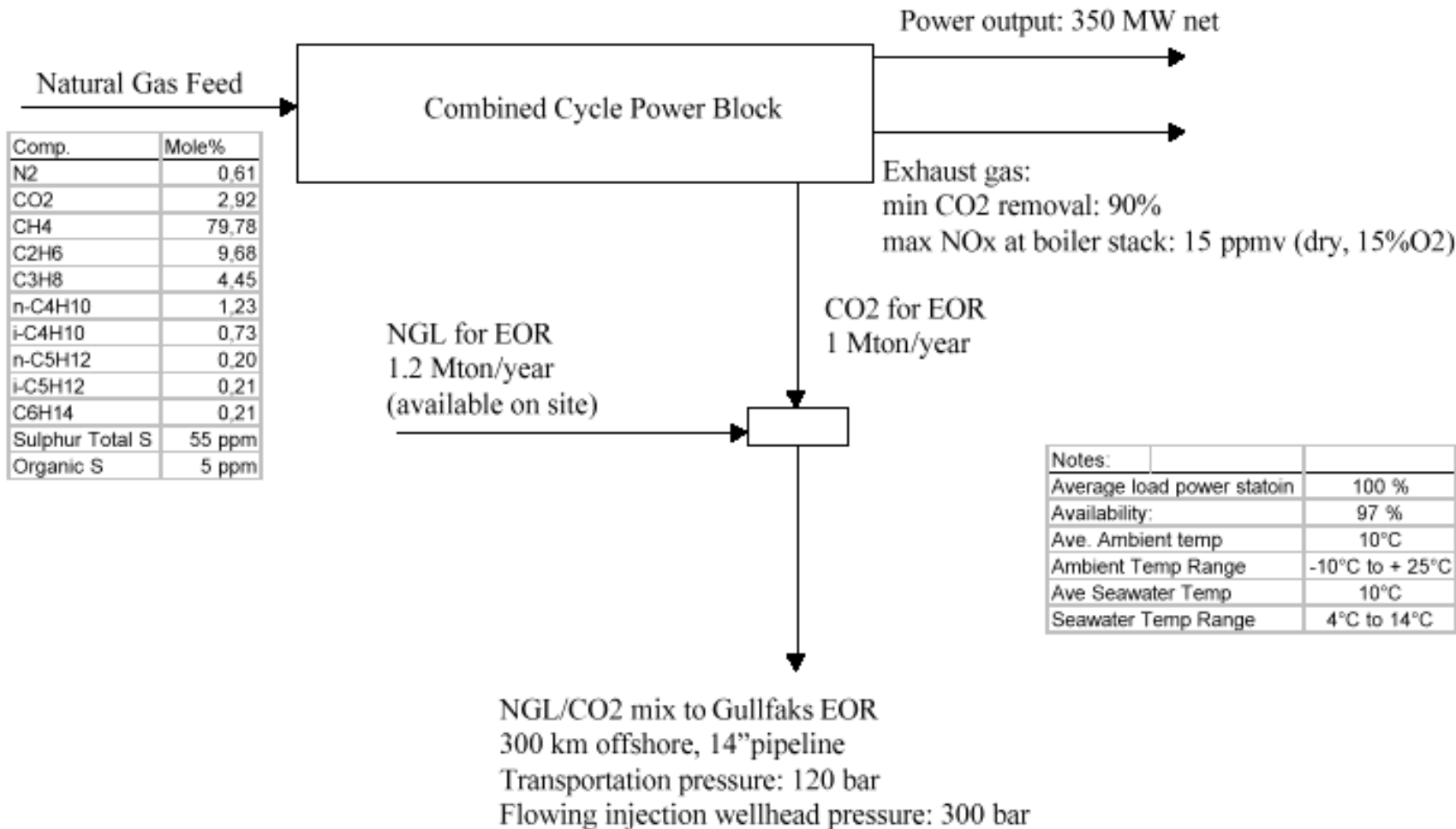


**Reforming:**  
 $H_2O : CH_4 = 3.5 / 2$   
 $CaO : CH_4 = 1.5$   
 $T = 600^\circ C$   
 $p = 5 \text{ bar}$

**Calcination:**  
 $T = 1000^\circ C$   
 $p = 5 \text{ bar}$   
 Atmosphere =  $CO_2$

## Batch process; circulation of gases





Notes:	
Average load power statoin	100 %
Availability:	97 %
Ave. Ambient temp	10°C
Ambient Temp Range	-10°C to + 25°C
Ave Seawater Temp	10°C
Seawater Temp Range	4°C to 14°C

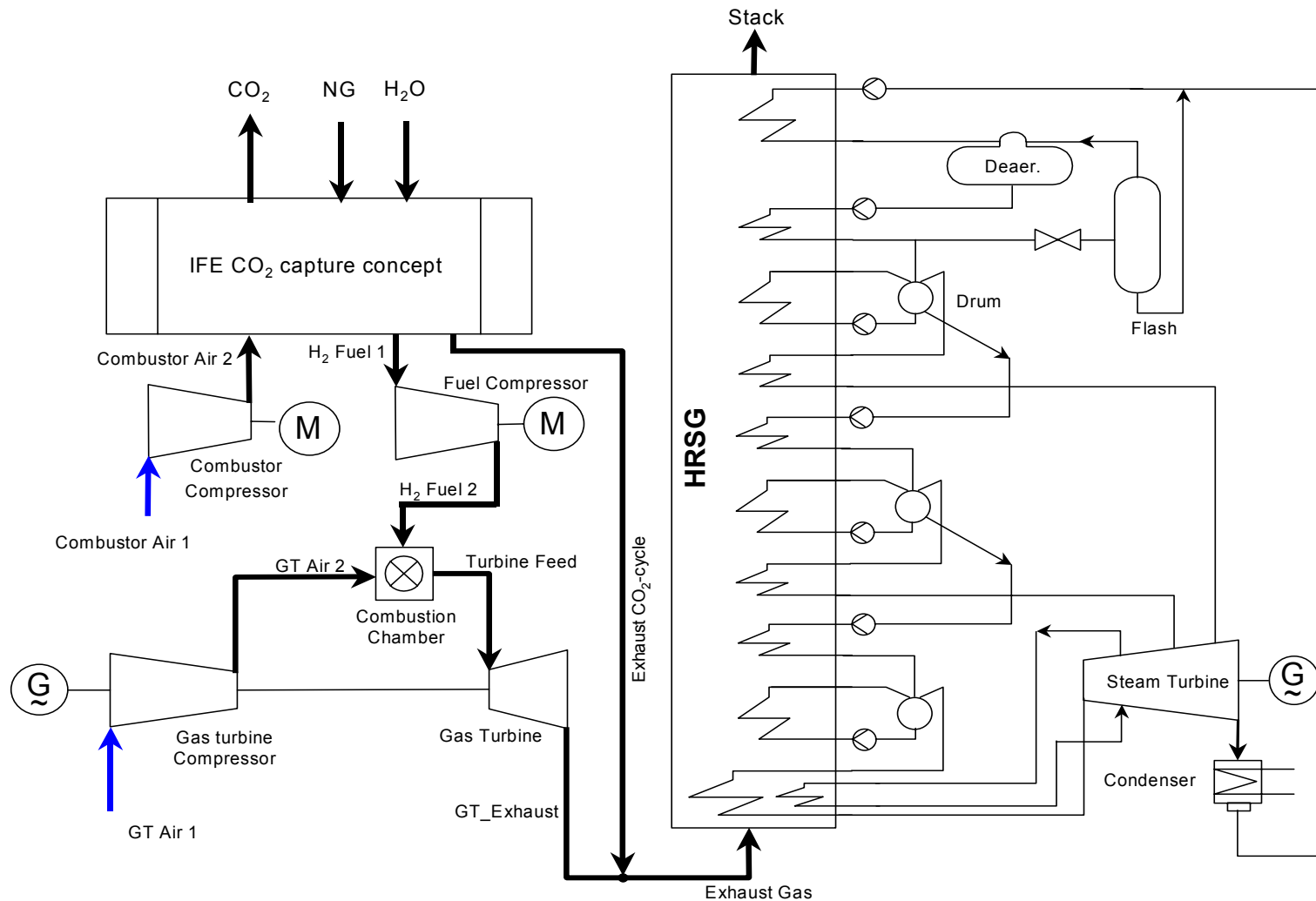
**Norwegian CO2 Capture  
and sequestration Scenario**

 **STATOIL** Rev. 1 January 24, 2001

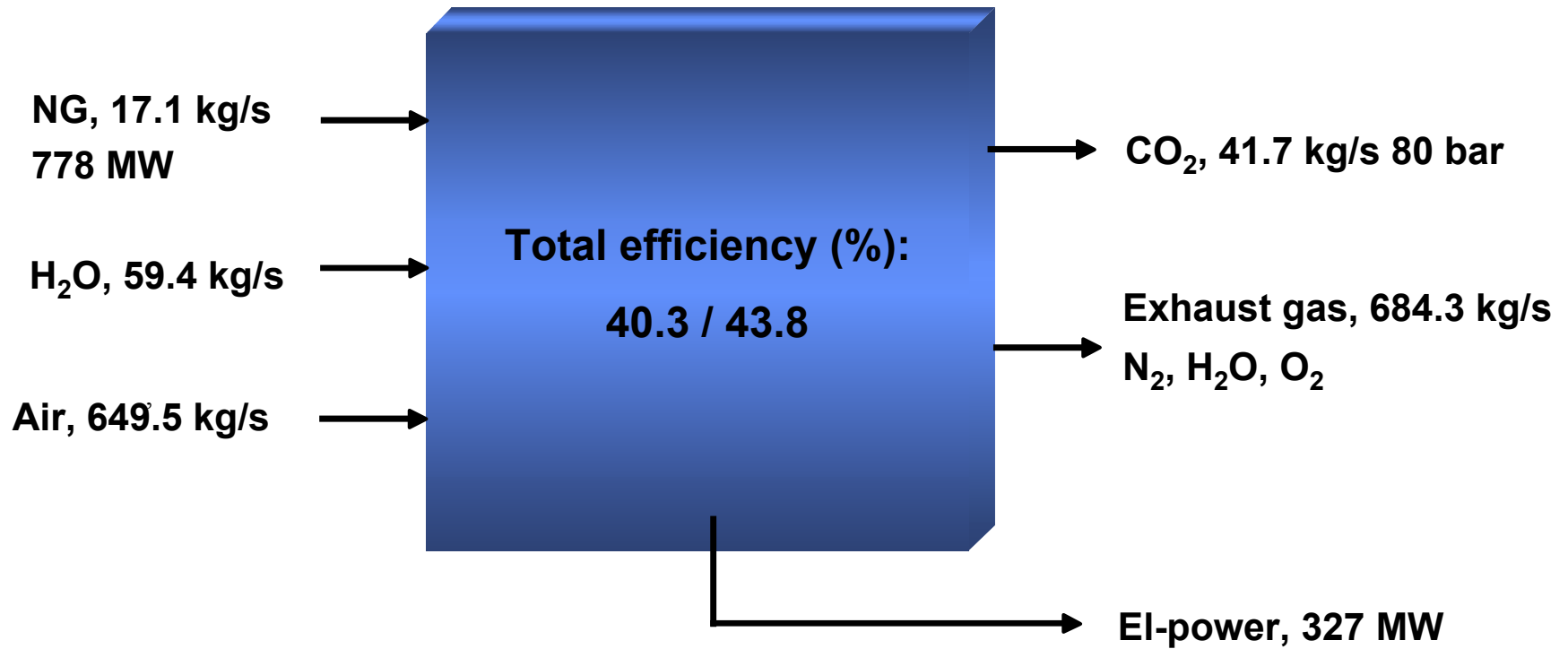




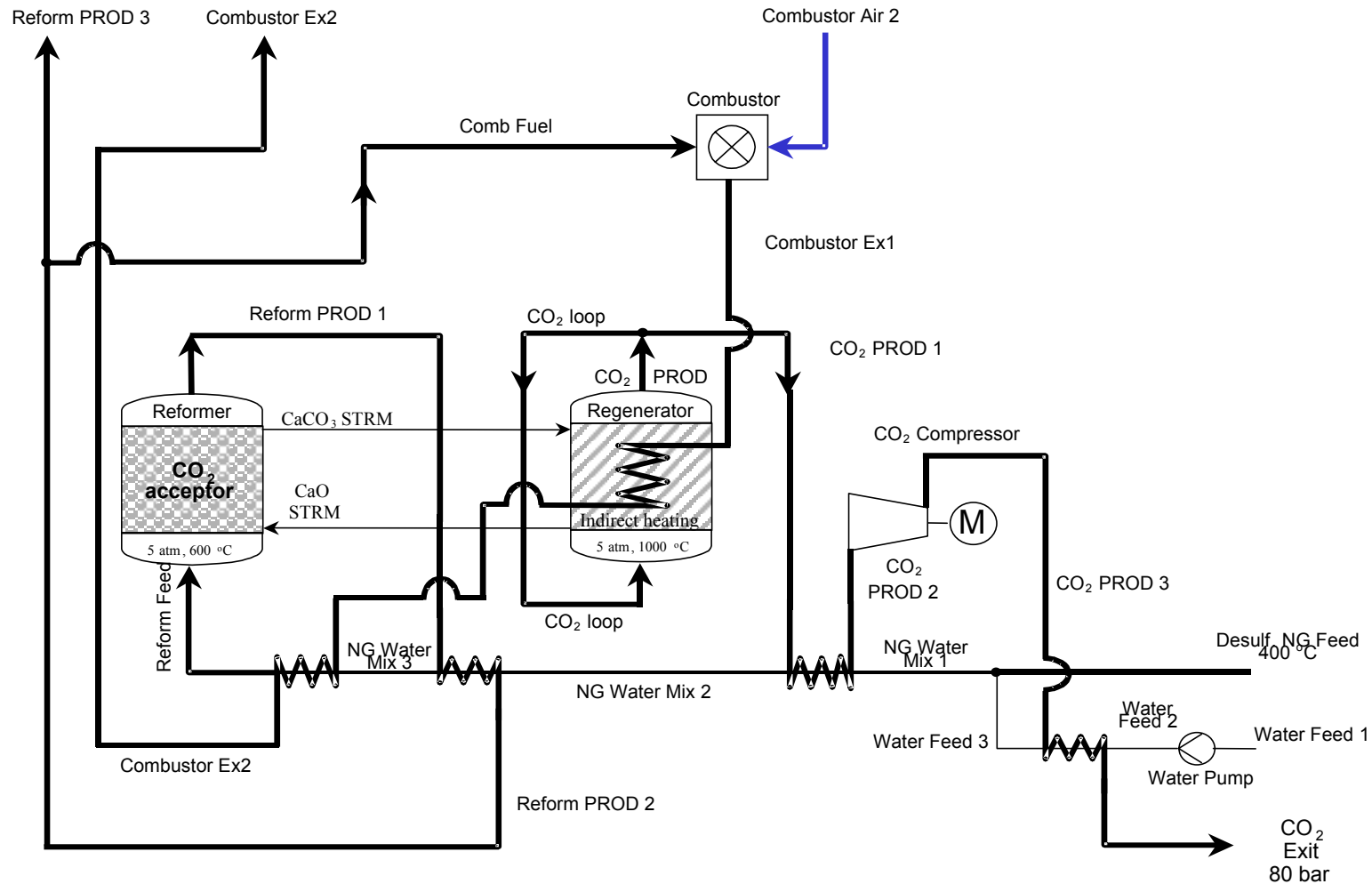
# Process flow diagram, CCBP-system



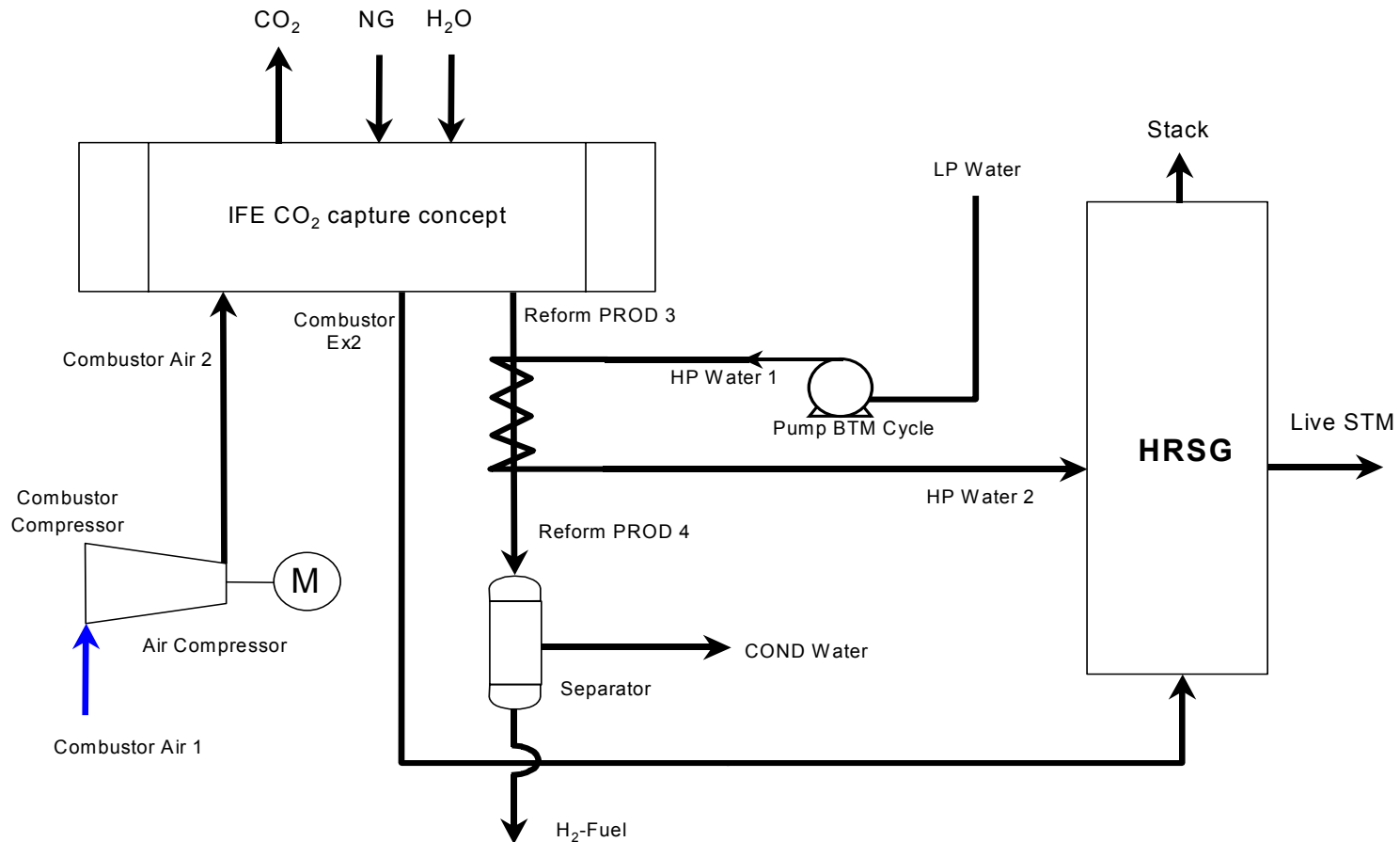
# Production and consumption figures, H<sub>2</sub>-production with integrated CO<sub>2</sub>- capture and CCPP-system



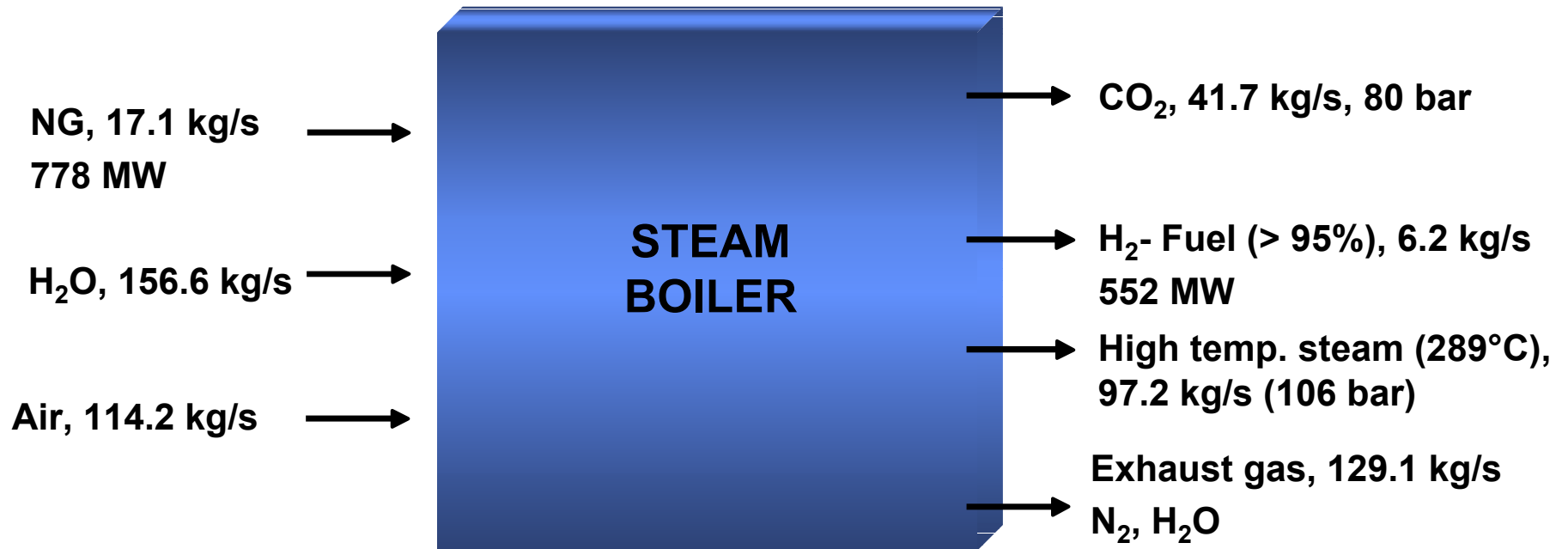
# IFE CO<sub>2</sub>-capture concept; steam boiler-system



# Process flow diagram, steam boiler-system



# Production and consumption figures, H<sub>2</sub>-production with integrated CO<sub>2</sub>- capture and steam boiler



# Assumptions and simplifications

- Heat from the desulphuriser is used for production of a natural gas feed at 400°C and 5.4 bar
- The conversion rate of hydrocarbons heavier than methane is 100 %
- The reformer conversion rate for methane is 93 %
- The CO<sub>2</sub>-adsorption rate is 95 %
- The CaCO<sub>3</sub> conversion in the calcination reaction is 99 %
- A catalyst is not defined in the simulation - the catalyst is not circulating between the reactors and does not affect the heat balance of the system
- The combustion chambers of the regenerator are inside the reaction vessel for maximum heat transfer
- The adiabatic compressor efficiencies are assumed to be 85 %
- Heat loss to the environment from the CO<sub>2</sub> removal process reactors and pipes etc. is assumed to be 9.3 % of the heat transferred in the heat exchangers.
- The LHV efficiency of the CCPP is 58 %
- It is assumed that water vapour in the turbine fuel does not affect the firing properties of the combustion chamber. In stead of 50 % dry N<sub>2</sub> the simulations allows for steam in the inert fraction as long as 50 % H<sub>2</sub> is maintained.

# Summary

- **90% CO<sub>2</sub> removal is possible**
- **CCPP with electrical efficiency 58% (LHV) is reduced to 40 - 44%**
  - IFE CO<sub>2</sub>-capture concept is intended to operate at lower pressures, H<sub>2</sub>-fuel has to be compressed
  - Need for sulphur removal
- **Producing H<sub>2</sub> for a steam boiler, waste heat is also generated**
  - Heat can be used for preheating the boiler