# Carbon Dioxide Capture for Storage in Deep Geologic Formations – Results from the CO<sub>2</sub> Capture Project

Capture and Separation of Carbon Dioxide from Combustion Sources

## Edited by

## David C. Thomas

Senior Technical Advisor Advanced Resources International, Inc. 4603 Clearwater Lane Naperville, IL, USA

## Volume 1



Amsterdam – Boston – Heidelberg – London – New York – Oxford Paris – San Diego – San Francisco – Singapore – Sydney – Tokyo

#### Elsevier Internet Homepage - http://www.elsevier.com

Consult the Elsevier homepage for full catalogue information on all books, major reference works, journals, electronic products and services.

#### Elsevier Titles of Related Interest

AN END TO GLOBAL WARMING

L.O. Williams

ISBN: 0-08-044045-2, 2002

FUNDAMENTALS AND TECHNOLOGY OF COMBUSTION

F. El-Mahallawy, S. El-Din Habik ISBN: 0-08-044106-8, 2002

GREENHOUSE GAS CONTROL TECHNOLOGIES: 6TH INTERNATIONAL CONFERENCE

John Gale, Yoichi Kaya ISBN: 0-08-044276-5, 2003

MITIGATING CLIMATE CHANGE: FLEXIBILITY MECHANISMS

T. Jackson

ISBN: 0-08-044092-4, 2001

#### **Related Journals:**

Elsevier publishes a wide-ranging portfolio of high quality research journals, encompassing the energy policy, environmental, and renewable energy fields. A sample journal issue is available online by visiting the Elsevier web site (details at the top of this page). Leading titles include:

Energy Policy
Renewable Energy
Energy Conversion and Management
Biomass & Bioenergy
Environmental Science & Policy
Global and Planetary Change
Atmospheric Environment
Chemosphere – Global Change Science
Fuel, Combustion & Flame
Fuel Processing Technology

All journals are available online via ScienceDirect: www.sciencedirect.com

#### To Contact the Publisher

Elsevier welcomes enquiries concerning publishing proposals: books, journal special issues, conference proceedings, etc. All formats and media can be considered. Should you have a publishing proposal you wish to discuss, please contact, without obligation, the publisher responsible for Elsevier's Energy program:

Henri van Dorssen

Publisher Elsevier Ltd

The Boulevard, Langford Lane Phone: +44 1865 84 3682
Kidlington, Oxford Fax: +44 1865 84 3931
OX5 1GB, UK E.mail: h.dorssen@elsevier.com

General enquiries, including placing orders, should be directed to Elsevier's Regional Sales Offices – please access the Elsevier homepage for full contact details (homepage details at the top of this page).

ELSEVIER B.V. Radarweg 29 P.O. Box 211, 1000 AE Amsterdam The Netherlands ELSEVIER Inc. 525 B Street, Suite 1900 San Diego, CA 92101-4495 USA ELSEVIER Ltd The Boulevard, Langford Lane Kidlington, Oxford OX5 1GB UK ELSEVIER Ltd 84 Theobalds Road London WC1X 8RR

© 2005 Elsevier Ltd. All rights reserved.

This work is protected under copyright by Elsevier Ltd, and the following terms and conditions apply to its use:

#### Photocopying

Single photocopies of single chapters may be made for personal use as allowed by national copyright laws. Permission of the Publisher and payment of a fee is required for all other photocopying, including multiple or systematic copying, copying for advertising or promotional purposes, resale, and all forms of document delivery. Special rates are available for educational institutions that wish to make photocopies for non-profit educational classroom use.

Permissions may be sought directly from Elsevier's Rights Department in Oxford, UK: phone (+44) 1865 843830, fax (+44) 1865 853333, e-mail: permissions@elsevier.com. Requests may also be completed on-line via the Elsevier homepage (http://www.elsevier.com/locate/permissions).

In the USA, users may clear permissions and make payments through the Copyright Clearance Center, Inc., 222 Rosewood Drive, Danvers, MA 01923, USA; phone: (+1) (978) 7508400, fax: (+1) (978) 7504744, and in the UK through the Copyright Licensing Agency Rapid Clearance Service (CLARCS), 90 Tottenham Court Road, London W1P 0LP, UK; phone: (+44) 20 7631 5555; fax: (+44) 20 7631 5500. Other countries may have a local reprographic rights agency for payments.

#### Derivative Works

Tables of contents may be reproduced for internal circulation, but permission of the Publisher is required for external resale or distribution of such material. Permission of the Publisher is required for all other derivative works, including compilations and translations.

#### Electronic Storage or Usage

Permission of the Publisher is required to store or use electronically any material contained in this work, including any chapter or part of a chapter.

Except as outlined above, no part of this work may be reproduced, stored in a retrieval system or transmitted in any form or by any means, electronic, mechanical, photocopying, recording or otherwise, without prior written permission of the Publisher.

Address permissions requests to: Elsevier's Rights Department, at the fax and e-mail addresses noted above.

#### Notice

No responsibility is assumed by the Publisher for any injury and/or damage to persons or property as a matter of products liability, negligence or otherwise, or from any use or operation of any methods, products, instructions or ideas contained in the material herein. Because of rapid advances in the medical sciences, in particular, independent verification of diagnoses and drug dosages should be made.

First edition 2005

Library of Congress Cataloging in Publication Data A catalog record is available from the Library of Congress.

British Library Cataloguing in Publication Data A catalogue record is available from the British Library.

ISBN: 0-08-044570-5 (2 volume set)

Volume 1: Chapters 8, 9, 13, 14, 16, 17, 18, 24 and 32 were written with support of the U.S. Department of Energy under Contract No. DE-FC26-01NT41145. The Government reserves for itself and others acting on its behalf a royalty-free, non-exclusive, irrevocable, worldwide license for Governmental purposes to publish, distribute, translate, duplicate, exhibit and perform these copyrighted papers. EU co-funded work appears in chapters 19, 20, 21, 22, 23, 33, 34, 35, 36 and 37. Norwegian Research Council (Klimatek) co-funded work appears in chapters 1, 5, 7, 10, 12, 15 and 32.

Volume 2: The Storage Preface, Storage Integrity Preface, Monitoring and Verification Preface, Risk Assessment Preface and Chapters 1, 4, 6, 8, 13, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29, 30, 31, 32, 33 were written with support of the U.S. Department of Energy under Contract No. DE-FC26-01NT41145. The Government reserves for itself and others acting on its behalf a royalty-free, non-exclusive, irrevocable, worldwide license for Governmental purposes to publish, distribute, translate, duplicate, exhibit and perform these copyrighted papers. Norwegian Research Council (Klimatek) co-funded work appears in chapters 9, 15 and 16.

The paper used in this publication meets the requirements of ANSI/NISO Z39.48-1992 (Permanence of Paper).
Printed in The Netherlands.



www.elsevier.com | www.bookaid.org | www.sabre.org

ELSEVIER

BOOK AID International

Sabre Foundation

## Chapter 11

# PRE-COMBUSTION DECARBONISATION TECHNOLOGY SUMMARY

Henrik Andersen Norsk Hydro, ASA, Oslo, Norway

#### **ABSTRACT**

The  $CO_2$  Capture Project (CCP) was formed in late 2000 and after a review and evaluation phase began actual technical development work near the end of 2001. Most of the technology providers had only 2 years to complete their work. Even then, significant progress and advances in several key areas were made. New insights on adoption of existing technology in the CCP industrial scenarios were achieved. The key results from the pre-combustion technology development projects are:

- Four new advanced technologies were developed to "proof-of-concept" with significant advancement in efficiency, cost and CO<sub>2</sub> capture compared to the best available capture technology.
- The four technologies showed cost reduction potential in the range from 30 to 60%, with the Hydrogen Membrane Reformer demonstrating the highest potential.
- Three of the new advanced technologies were developed for different CCP scenarios. The designs were
  checked, integrated, and cost estimated by an independent contractor (Fluor) in order to assure design
  quality and consistency when comparing with the baseline technology, thus enhancing credibility of the
  conclusions.
- Significant advancements were made in hydrogen membrane materials covering a wide temperature range.
- Further development is needed to advance the most promising technologies, however, it is expected that new
  technologies can be developed and demonstrated in 2010–2015 with costs in the range of \$15–40 MM.
- Pre-combustion technology can be developed to meet stringent requirements on NO<sub>x</sub>, CO, and SO<sub>x</sub> formation. The lowest NO<sub>x</sub> formation was predicted to be 5 ppm vol. from a combined cycle gas turbine. For open-cycle gas turbines, the NO<sub>x</sub> formation was reduced by 50%. CO and SO<sub>x</sub> formation were virtually zero.
- Pre-combustion technology can be designed as stand-alone facilities for both retrofit and new build
  applications giving a wide application range and benefits with respect to integration in existing complex
  facilities, e.g. refineries.
- Pre-combustion technology can be used for other applications, e.g. gas-to-liquids (GTL), ammonia, hydrogen
  and syngas production, thus increasing the economic potential of the technology and return of investment.
- Significant improvement in energy and CO<sub>2</sub> capture efficiency was obtained for several technologies, resulting in an efficiency penalty for combined cycle gas turbines of less than 5% with nearly 100% CO<sub>2</sub> capture.
- A 15% improvement of gas turbine heat rate can be obtained when switching from natural gas to hydrogenrich fuel, making the pre-combustion technology a strong candidate for the large numbers of open-cycle gas turbines in operation in the US.
- Demonstrated very low CO<sub>2</sub> avoided costs for the Canadian scenario—CO<sub>2</sub> capture from petroleum coke fired IGCC—approximately \$10-15 per ton.
- Existing pre-combustion technology can be considered proven for a wide range of CO<sub>2</sub> capture applications including the CCP scenarios.

#### INTRODUCTION

The CO<sub>2</sub> Capture Project (CCP) pre-combustion technology development was the largest CO<sub>2</sub> capture program in the CCP. It involved 13 individual projects completed by 20 different technology suppliers.

The studies are divided into three key areas:

- Membrane studies: development of hydrogen membranes and reactors for steam reforming and water gas shift application.
- Enhanced hydrogen production: novel non-membrane technology for syngas and CO<sub>2</sub> capture application.
- Integration and scale-up studies: existing technology integrated and optimised for the CCP scenarios.

All the technologies have been developed to be used in the real-life industrial scenarios defined by the CCP. This approach gave the most insight into the economic potential and technical performance of the technologies.

## History of Pre-combustion Technology

Pre-combustion technology is based on well-known technologies that are currently used in commercial operations such as: hydrogen, ammonia and syngas production. The technology comprises two main steps:

reforming/conversion of fossil fuel to syngas (a mixture containing hydrogen, CO, and  $CO_2$ ), and separation of  $CO_2$  and hydrogen to produce a hydrogen-rich stream.

Conversion of fossil fuel to syngas dates back to several centuries when coal was the primary energy source. The Scottish engineer William Murdoch was the first to convert fossil fuel to syngas who in 1792 used the syngas to light his house. Syngas was later called "town gas" or "city gas". "Gas" lighting was widespread between 1800 and 1920 for lighting homes and businesses. In the United States, more than 1000 town gas plants (Figure 1) were in operation in 1905. The technology developed from gasification of coal to reforming of natural gas through use of catalysts. Steam reforming technology, introduced in the 1930s, remains the primary method to convert natural gas to syngas. More than 90% of the present hydrogen production—500 billion m³ per year according to IEA—is based on reforming of fossil fuel. This volume suggests that about 500 reformers with an average capacity of 100,000 Nm³/h hydrogen are in operation worldwide.



Figure 1: Town Gas Plant from 1911. Producer-Gas Plant, St. Louis, Missouri, ca 1911. Source: Fernald and Smith/US Bureau of Mines (1911).

Reforming technology development for natural gas and similar fossil fuels proceeded along two technical lines:

- steam methane reforming, an endothermic process that requires heat addition to convert a mixture of steam and natural gas to syngas at high temperature, and
- autothermal reforming (ATR), an exothermic process to convert a mixture of steam, natural gas and oxygen into syngas and excess heat.

Improvements in steam methane reforming since its introduction have been made through increasing operating pressure and temperature by development of new catalysts and reactor materials.

Combining the two steps of syngas production and separation of hydrogen and  $CO_2$  is a well-established technology mainly used in production of syngas for ammonia production. The first system was introduced in the 1940s. It used low-pressure steam reforming followed by compression to 15 barg and  $CO_2$  separation from the  $H_2$  through an amine separation process using 20% monoethanolamine (MEA). In the mid-1950s, a separation technology using hot potassium carbonate was introduced and in the late 1970s, activated MDEA solvent was introduced leading to a significant improvement in energy efficiency.

The largest ammonia plants (Figure 2) produce about 2000-2200 ammonia t/day, which requires hydrogen production of about  $150,000-200,000 \text{ Nm}^3\text{/h}$  or 450-600 MW (LHV). Approximately 1,000,000 tpa of  $CO_2$  is captured in the largest plants and compressed to 160 barg for use in urea production.

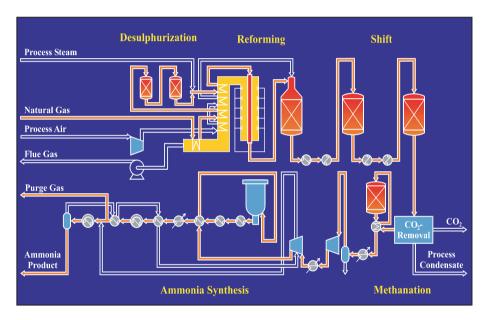
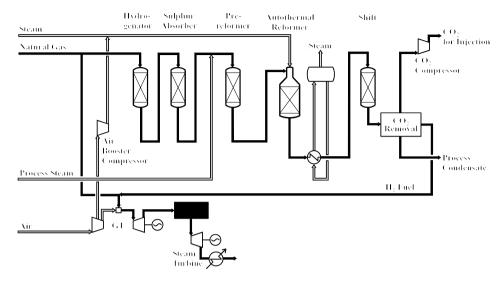


Figure 2: Ammonia process scheme. Source: Haldor Topsoe.

#### State-of-the-Art Pre-combustion CO<sub>2</sub> Capture Technology

One of the first attempts to develop a pre-combustion CO<sub>2</sub> capture process was in 1998, when Norsk Hydro launched the *Hydrokraft* project (Figure 3). The concept was based on air-blown ATR technology to produce a mixture of hydrogen and nitrogen for use as fuel for a large gas turbine. The project gave important insights into pre-combustion technology and into the pre-combustion base line technology.

Pre-combustion technology is a very complex process involving a number of catalytic steps, heating to high temperatures and cooling to low temperatures. Improvements that favor one part of the process might be a disadvantage for another part. As an example, reforming is favored by low pressure, but separation of CO<sub>2</sub> is favored by high pressure. Energy efficiency is favored by low steam addition but hydrogen production is favored by high steam addition. The energy loss is significant—almost 25% of the energy is lost in present pre-combustion processes. Further, the investment required to build a pre-combustion plant is tremendous. A combined cycle gas turbine power plant using pre-combustion processes to make hydrogen fuel will be double the cost of the same facility fueled with natural gas.



**Figure 3:** The "Hydrokraft" concept or Integrated Reforming Combined Cycle (IRCC)—Pre combustion baseline technology. Source: Haldor Topsoe/Hydro.

#### Technology Program Development

The pre-combustion team used their understanding of historical hydrogen production of other current "State-of-the-art" technology to develop two approaches for improving the pre-combustion technology:

system optimization by use of well-known technologies, and new technologies based on advanced separation combining both reaction and separation.

Since different technologies are at different levels of development and different risk factors are associated to the technologies, a key challenge was to define a program with a balanced portfolio—taking into account that success was a result of the balance between risk and potential benefits. The team agreed to invest in technology with less potential and lower risk but with a higher chance for success.

As mentioned previously, the core technologies of pre-combustion technology have been developed for commercial markets for more than 50 years. This has created a large industry and a core area in many universities and institutes. The challenge for the team was to identify the best players in the market and to engage them in the program.

During the bidding phase, all the technology providers were requested to define a program scope that would bring the technology to a "proof-of-concept" stage. However, in order to monitor progress and differences in timing and be able to (re)-direct the program, a stage gate process was adopted. Each project identified critical milestones, e.g. material durability and performance, for different stages in the development work—called stage gate criteria. The definition of stage gate criteria was established in collaboration between the CCP and the technology suppliers. The stage gate process made decision taking and choices easier.

#### PRE-COMBUSTION TECHNOLOGY DEVELOPMENT PROGRAM

Based on the review and evaluation of technology and working with potential technology suppliers, the program outlined in Table 1 was defined in 2001.

TABLE 1 PRE-COMBUSTION TECHNOLOGY STUDIES STATUS: JAN 2004

References	Project title	Co funder	Technology provider	Status
1.2.1.1	Sulfur-tolerant membrane study	DoE	Fluor, SOFCo, Eltron, TDA Research, CSM, ECN, University of Cincinatti	Did not pass complete stage gate review in April '03. Entered into phase II with reduced and revised scope. Eltron, Fluor and SOFCo were remaining technology providers. Project completed.
1.2.1.2	Hydrogen membrane reactor	EU	BP, Norsk Hydro, KTH, Sintef, University of Twente, Institute for membrane technology, University of Zaragoza	Passed stage gate review. Project completed
1.2.1.3	Hydrogen membrane reformer	Klimatek	Norsk Hydro	Passed stage gate project completed with successful "proof-of-concept" test
1.2.1.4	Precombustion membrane reactor study	CCP	Haldor Topsoe	Completed in Feb. 2001
1.2.2.1	Advanced technology for separation and capture of CO <sub>2</sub> from gasifier process producing electrical power, steam and hydrogen	DoE	Fluor Federal	Completed Oct 2003
1.2.3.1	Study of gas turbine retrofit requirements to burn decarbo- nised fuel (hydrogen)	DoE	General Electric	Completed Dec 2003
1.2.3.2 1.2.3.3	Standardized PCDC Very large-scale autothermal	Klimatak CCP	Jacobs Jacobs	Completed Dec 2003 Completed in May 2003
1.2.3.4	reforming Advanced	CCP	Foster-Wheeler	Completed in
1.2.3.5	syngas study Compact reformer with advanced pressure swing adsorption system for hydrogen fuel production	DoE	Davy/APCI	Feb 2001 Compact reformer dropped. Advanced PSA study completed Dec 2003

(continued)

TABLE 1 CONTINUED

References	Project title	Co funder	Technology provider	Status
1.2.4.1	Capture study integrated reports	DoE	ARI	Completed
1.2.5.1	Generation of H <sub>2</sub> fuels	Klimatak	IFE	Completed Feb 2002
1.2.6.1	Production of hydrogen fuel by sorbent-enhanced water gas shift reaction	DoE	Air Products and Chemicals	Passed phase II stage gate review. Phase III completed with "proof-of-concept" test

The technology development program was carried out over the 2002-2003 timeframe and is summarized here and detailed in the following chapters.

The results indicate that the technologies in the membrane area advanced much more than anticipated given the short time (16-24 months) for actual development work. The risk element was high with a reasonable likelihood of failure. Even so the membrane developers have overcome significant barriers and are well positioned to continue the work.

## Sulfur-Tolerant Membrane Study (Table 1, Ref. 1.2.1.1)

The study objective was to develop a sulfur-tolerant membrane operating at water gas shift conditions. Four membrane developers were given 1 year to develop a membrane with significant flux and selectivity for a sour syngas. None of the developers reached the target; however, a promising membrane for sweet syngas condition was identified. The pre-combustion team agreed to re-direct the program and continue the development for sweet syngas application. The program continued with Eltron as the membrane developer, SOFCo as the commercial reactor designer, and Fluor as responsible for process integration.

The development of a novel low-cost compact design for a membrane water gas shift reactor and improved membrane for a water gas shift reactor with selectivity and flux was achieved. This technology will lead to reduced reactor and membrane costs in the US DOE refinery scenario and the technology shows a potential of 30–35% reduction in CO<sub>2</sub> avoided cost when using refinery off-gas.

The team considers this technology promising with medium potential and medium risk. They recommend that the work be continued and to, also continue searching for a sulfur-tolerant hydrogen membrane.

It should be noted that a concept based on gasification of heavy fuel oil was also developed. The CO<sub>2</sub>-avoided cost was higher than the amine post-combustion baseline technology and was not pursued further.

#### Hydrogen Membrane Reactor Technology (Table 1, Ref. 1.2.1.2)

A consortium of European membrane developers was created with a common task of developing novel hydrogen membranes that could be used in pre-combustion applications—the CCP EU refinery scenario. The membrane types were ultra-thin Pd-membranes, silica-based ceramic membranes, and Pd-coated zeolites. The most promising membrane was the dense Pd/Ag membrane in which a 1  $\mu$ m thick film was manufactured by a method developed by SINTEF and deposited on a porous stainless steel support tube. Significant advancement was also achieved in the silica-based ceramic membrane where a selectivity of 1000 was obtained.

A reactor concept incorporating the Pd-membrane was developed with an associated process scheme for production of a hydrogen fuel mixture for heaters and boilers. The technology demonstrates high energy and CO<sub>2</sub> capture efficiency and low cost. A CO<sub>2</sub> reduction cost of 25–30% using refinery off-gas was achieved.

The team has recommended that work on the Pd-membrane with a focus on long-term testing of stability and performance be continued.

### Hydrogen Membrane Reactor Technology (Table 1, Ref. 1.2.1.3)

Norsk Hydro was selected to develop a technology based on high-temperature ceramic hydrogen membranes for combined cycle gas turbines as part of the Klimatek-funded NORCAP project. The technology principle is similar to some of the concepts being studied by the Oxyfuel team for oxygen-conducting ceramic membranes. The first phase of the project aimed at developing a membrane that could achieve significant flux in order to meet the CCP targets. This work was done in collaboration with the University of Oslo and SINTEF. At the end of Phase 1, a membrane was synthesized with sufficient flux. The membrane reformer system showed untouchable performance in the NORCAP Norwegian scenario with very high energy efficiency (approximately 90–91% LHV), low NO<sub>x</sub> formation—5 ppm vol. and a potential CO<sub>2</sub>-avoided cost reduction of 50–55%. Proof-of-concept tests confirmed hydrogen flux above expectations. The team recommended continued work on the technology in the extended Klimatek program for 2004.

## Advanced Technology for Separation and Capture of $CO_2$ from Gasification, Producing Electrical Power, Steam and Hydrogen (Table 1, Ref. 1.2.2.1)

Fluor completed a complete study of pre-combustion technologies for a petroleum coke fired IGCC with production of steam, electricity and hydrogen. Uncontrolled and baseline cases were established, several pre-combustion technologies were screened, and one technology was selected for detailed design and costing. The results showed that with conventional technology, a  $CO_2$  avoided cost of approximately \$15 per ton could be obtained. This gives very little room for improvement. Screening of different pre-combustion options was based on multiple criteria, e.g.  $CO_2$  recovery above 85%, hydrogen delivery at gas turbine pressure, sulfur tolerance, sulfur content in  $CO_2$  stream and so on. Using these criteria, the Fluor  $CO_2LDSEP$  was seen as the most suitable option. Due to confidentiality issues, the capital cost was assessed by a sensitivity analysis—showing that the  $CO_2$  avoided cost for the technology was in the range \$10–20 per ton.

Results indicate that very low CO<sub>2</sub> avoided cost can be obtained in US DOE Canadian scenario by adopting pre-combustion technology—in the range \$10–15 per ton CO<sub>2</sub>.

#### Gas Turbine Retrofit Requirements to Burn Decarbonised Fuel (Hydrogen) (Table 1, Ref. 1.2.3.1)

One of the critical success factors for pre-combustion technology is that hydrogen-rich fuel can be used in multiple combustion processes. Hydrogen-fuel for use in gas turbines' combustors is an area that requires special attention in terms of performance, lifetime, and cost. The leading gas turbine supplier for syngas fuel combustors was selected to conduct the study—General Electric. The study results were very encouraging and, in particular, retrofit of gas turbines was confirmed to be feasible. In addition, an improved heat rate of 15% was estimated which will reduce the size of the pre-combustion plant and increase overall energy efficiency for a projected power plant. Changing from natural gas to hydrogen-rich fuel, GE estimated that a 50% NO $_x$  reduction could be achieved and for some hydrogen fuel mixtures single-digit NO $_x$  ppm levels can be obtained.

#### Standardized Pre-combustion De-carbonisation (PCDC) Technology (Table 1, Ref. 1.2.3.2)

The pre-combustion team initiated a study to evaluate cost-reducing options for pre-combustion baseline technologies. The focus was on cost savings from repeat design, modularization, mechanical codes, pre-fabrication and so on. The results were somewhat disappointing only demonstrating cost savings in the order 15–20%. Further work in this area should focus on rotating equipment, which contributed 60% to the total installed cost.

#### Very Large-Scale Autothermal Reforming (Table 1, Ref. 1.2.3.3)

A key feature of pre-combustion technology is the potential of designing very large capture plants in a central location with distribution of the hydrogen fuel to combustion operations thus obtaining the benefits from economy-of-scale. The study confirmed that a pre-combustion technology could be built for the Alaska scenario—capturing over 2 million tpa of CO<sub>2</sub> and producing more than 750 MW of fuel. The team felt, however, that the proposed process design was not optimal for the Alaska scenario and further work

would be needed. Economic modeling estimates showed that the CO<sub>2</sub> reduction potential was less than 15%. However, looking at the option at a similar maturity as the post-combustion baseline technology—one conclusion from the work could be that pre-combustion is preferred over post-combustion technology.

#### Advanced Syngas Study (Table 1, Ref. 1.2.3.4)

Several technologies that are commercially available or close to commercialization were studied as the baseline technology for the Norwegian scenario—Integrated Reforming Combined Cycle technologies. The study results demonstrated limited potential whether for adopting new technology or for optimizing the concepts. However, the study showed that a pre-combustion baseline was lower cost technology than post-combustion but had higher energy consumption.

## Compact Reformer with Advanced Pressure Swing Adsorption System for Hydrogen Fuel Production (Table 1, Ref. 1.2.3.5)

The scope of the work was reduced to evaluate only the advanced pressure swing adsorption system since the compact reformer developer (Davy) would not agree to the needed contract terms to allow integrated analysis. Results showed that pressure swing adsorption cycles that couple hydrogen purification with carbon dioxide recovery system offer higher hydrogen recovery with the same number of adsorbent columns. It was determined that a single-train adsorption system can provide 0.8 million tpa of CO<sub>2</sub> at up to 99.7% purity and with a CO<sub>2</sub> recovery of up to 93%. The economics of the technology and integration in a complete pre-combustion scheme is recommended.

#### Generation of H<sub>2</sub> Fuels (Table 1, Ref. 1.2.5.1)

A process scheme using CaO as an oxidant to drive the reforming reaction was developed for the Norwegian and EU refinery scenarios. The project goal was to estimate CO<sub>2</sub> capture and energy efficiency to be expected from the technology. Results showed that for combined cycle gas turbines, the technology could not obtain satisfactory energy efficiency reaching only 40% (LHV). For heater and boiler options, energy efficiency was estimated to be approximately 82% (LHV) if an electricity credit is included. A CO<sub>2</sub> capture efficiency of 90% could be obtained for both applications. CaO reacts to form CaCO<sub>3</sub> and must be converted back to CaO by calcinations for recycle to the reforming reactor. The key challenge is to develop a reactor system that can recycle solid materials efficiently. The technical risk associated with development of the technology must be considered high and fundamental studies and lab testing must be conducted before pilot testing can be realized. The recommendation from the team is to study the refinery case in more detail and establish a cost estimate for the process before considering laboratory development work.

Production of Hydrogen Fuel by Sorbent-Enhanced Water Gas Shift Reaction (Table 1, Ref. 1.2.6.1) The leading adsorbent material ADS1-2 has a CO<sub>2</sub> removal capacity of up to 1.1% in PDU cyclic testing. A new adsorbent with the potential for significantly higher CO<sub>2</sub> capacities than other adsorbents has been identified. Further development of this adsorbent could lead to significant improvement of the sorption-enhanced water gas shift reactor scheme for gas turbine applications like the Alaska scenario or the Norwegian Scenario. The technology demonstrated significant cost reduction—in the range 30–35%—compared to the baseline technology. The technology is considered to be at a more mature stage than membrane technologies and has high potential. The team recommends continuing work on this technology.

#### CONCLUSIONS

#### Pre-combustion Technology Application to the CCP Scenarios

A key advantage of pre-combustion technology is its fuel flexibility and ability to convert all types of fossil fuels into syngas. That alone makes pre-combustion technology the only capture technology applicable to all of the CCP scenarios. Another advantage of the pre-combustion technology is that hydrogen fuel production and CO<sub>2</sub> capture take place in a dedicated plant at a central location yielding significant economy of scale compared to the other capture technologies. Each CCP scenario includes retrofit cases. There was concern that pre-combustion technology could not be retrofit to gas turbines. Our studies show that turbines can be retrofit to burn hydrogen fuels (GE study).

Three of the most promising technologies were applied to CCP scenarios using technology provider information:

- Hydrogen Membrane Reformer for NorCap scenario.
- · Membrane water gas shift reactor for the UK refinery.
- Sorption-enhanced water gas shift reactor for the Alaska scenario.

Fluor evaluated the technologies in the subject scenarios. Their studies included: the integrated design, quality assurance, and cost estimation. The study created a unique platform for comparison against the baseline technologies thus giving new insights on how the technologies performed in the given scenario and credibility to the cost reduction potential estimated by the CEM team.

#### **Economics**

Economic modeling results discussed in Chapter 4 of this Volume showed that three novel pre-combustion technologies had a significant potential to reduce  $CO_2$  avoided cost compared to the baseline technology. The Hydrogen Membrane Reformer, assessed in the NorCap scenario, demonstrated a  $CO_2$  avoided cost reduction potential of 60% compared to the baseline technology. This is equal to an annual saving of approximately \$35 million for a 400 MW power plant.

The membrane water gas shift system was assessed in the Canadian gasification scenario for two different fuels (refinery fuel oil and refinery off-gas). The results were remarkable. In the refinery fuel oil case, the  $\rm CO_2$  avoided cost increased when compared to the baseline technology because both energy loss and capital cost required to gasify the fuel oil was much higher than anticipated. In the refinery off-gas case, the much more efficient reforming process, ATR, could be used. For that case, the  $\rm CO_2$  avoided cost was reduced by  $\rm 30{\text -}40\%$  compared to the baseline technology.

Sorption-enhanced water gas shift was assessed in the NorCap and Alaska scenarios. In the NorCap scenario up to 44% cost reduction was achieved by using air-blown ATR, but for the Alaska scenario only 19% cost reduction was achieved. One important remark is that the improved gas turbine efficiency gained by switching from natural gas to a hydrogen/nitrogen fuel-mix was not taken into account in the Alaska scenario. Further, the Alaska scenario requires a very special design due to the extreme climatic conditions and location—this issue reduces the relative improvement when using new technologies.

#### Commercial Value—Present and Future

Pre-combustion technology for CO<sub>2</sub> capture accommodates a broader potential than the other capture technologies. The technology is widely applicable in any syngas production process such as: methanol, synfuel, ammonia, and hydrogen. Thus technology improvements made by the CCP can be adopted as well in these areas. As an example, a large GTL plant costs about \$1 billion and 60% of the cost relates to the syngas technology.

Significant improvement in hydrogen production technologies could be the base for low-cost hydrogen for future fuel cell vehicles. Hydrogen production with capture and storage of CO<sub>2</sub> will "bridge-the-gap" towards the renewable hydrogen economy.

#### **ACKNOWLEDGEMENTS**

The success of the pre-combustion team is the result of the committed, dedicated and inspiring involvement of key experts from the CCP partners: Peter Middleton, Mike Wilkinson and Paul Hurst from BP; Cliff Lowe, Karl Gerdes and Tom Leininger from Chevron Texaco; Jan Assink and Tom Mikus from Shell; Siv Aasland from Statoil; Martin Holysh from Suncor; Gabriele Clerice from ENI; Jan Schelling and Eline Jossang from Hydro. Without their hard work we never would have succeeded!